DITERPENES FROM MIKANIA SPECIES*

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Abstract—The investigation of four further *Mikania* species afforded several diterpenes, four geranylgeraniol derivatives and three kaurane epoxides. The structures were elucidated by spectroscopic methods and a few chemical transformations.

INTRODUCTION

The large genus Mikania (Compositae, tribe Eupatorieae) is mainly distributed throughout tropical America and placed in the subtribe Mikaniinae [1] as the only genus. So far about ten species have already been investigated chemically. While from some species highly oxygenated sesquiterpene lactones were reported [2–8], others afforded diterpenes, mainly ent-kaurene derivatives [3,8,9]. Also several flavones were isolated [10,11]. We now have investigated some more species, all from northeastern Brazil.

RESULTS AND DISCUSSION

The aerial parts of M. officinalis Mart. afforded germacrene D, γ -humulene, squalene, phytol, geranyl linalol (1), 22 and a further geranyl nerol derivative, the trihydroxy-aldehyde 2. The structure of 2 followed from the 1H NMR data and from those of the corresponding tetra-aldehyde 3 (Table 1). The stereochemistry of the 2,3-double bond could be assigned from the chemical shifts of H-1 and H-20 in the spectrum of 3, while the position of the two additional aldehyde groups followed from the shift differences in the spectra of 2 and 3 and from the signals of H-14 in that of 2. Spin decoupling further allowed the assignment of the H-5, H-9, H-12 and H-13 signals in the spectrum of 3. From the chemical shifts of H-5 and H-9 the position of the last aldehyde group could be assigned.

The roots afforded γ -humulene and the thymol derivatives 22–24. The aerial parts of M. sessilifolia DC. afforded germacrene D, α -humulene, squalene, the ent-kaurene derivatives 8 [13], 12 [14] and 13 [14] and four further diterpenes, the aldehydes 4–6 and the furan 7. Again the structures could be deduced from the ¹H NMR data (Table 1). 4 and 5 obviously were 2,3-E/Z-isomers as

could be deduced from the different chemical shifts of H-20. 4 has already been prepared from geranylgeraniol [15]. The stereochemistry of 6 and the position of the hydroxyl group also followed from the ¹H NMR data and the fragmentation pattern in the mass spectrum of 6. The H-2 signal was a doublet doublet quartet, indicating the presence of only one allylic proton in addition to the H-20 protons. Furthermore, the usually downfield shifted H-4 signals were missing. In the mass spectrum a fragment at m/z 100 (C₅H₈O₂) also was in agreement with a 4hydroxyl derivative of 4. The stereochemistry of the 2,3double bond followed from the chemical shift of H-20, while that of the other double bonds was assigned by comparing the corresponding shifts with those of geranylgeraniol and similar compounds. The structure of 7 clearly followed directly from the ¹H NMR data. The presence of a 2,3-disubstituted furan could be easily deduced from the signals of the two furan protons, their chemical shifts and couplings showed the presence of an αand a vicinal β -proton. We have named 7 mikanifuran.

The roots afforded **8**, **9** [16], **10** [17], **11** [18], **12** [14], **14** [19] and small amounts of further kaurane derivatives, the epoxides **17a**, **18a** and **19**, the first two only isolated as their methyl esters (**17b** and **18b**). To establish the relative stereochemistry of **17a** and **18a** as well as that of **19**, we have prepared the epoxides **20a** [20] and **20b** from *ent*-kaurene. ¹H NMR investigations using shift reagent allowed the assignment of the H-15 signals, which, however, could only be assigned in the β -epoxide **20a**. However, the chemical shifts of H-17 and $J_{17,17}$ allowed the assignment of the stereochemistry in all epoxides (Table 2).

The roots of M. luetzelburgii Mattf. afforded germacrene D, 8, 10, 12, 13 [14], 15 [15], 16 [21], 17a and 21, while the aerial parts gave germacrene D, lupeyl acetate 8, 10, 12, 13, 15, 16 and 21. The aerial parts of M. belemii K. et R. afforded α -humulene, caryophyllene and its 5,6-epoxide, gurjunene, 8, 10, 12, 13 and 16.

The results of the investigation of four further Mikania species show again that the chemistry of this large genus is not very uniform, though taxonomically it is one of the most distinctive genera in the Compositae. As in other

^{*} Part 360 in the series "Naturally Occurring Terpene Derivatives". For Part 359 see Bohlmann, F., Kramp, W., Grenz, M., Robinson, H. and King, R. M. (1981) Phytochemistry 20, 1907.

	2	3	4	5	6	7
H-1	4.16 d	10.27 d	10.00 d	9.92 d	10.07 d	7.20 d
H-2	5.70 br t	6.64 br d	5.90 br d	5.88 br d	6.13 ddq	6.15 d
H-4 }	2.15 m	2.88 br t	2.24 br s	2.60 br t	4.15 dd	_
H-5		2.77 br dt	_	2.25 br dt	2.42 br ddd 2.30 br ddd	3.28 br d
H-6	5.30 br t	6.39 br t	5.10 br t	5.09 br t	5.12 br t	5.26 br t
H-8 H-9	2.15 m	$\begin{array}{c} 2.19 \ m \\ 2.07 \ br \ dt \end{array}$	2.05 m	2.04 m	2.08 m	2.09 m
H-10	5.14 br t	5.13 br t	5.10 br t	5.09 br t	5.09 br t	5.10 br t
H-12 H-13	2.20 br t 2.45 br dt	$ \begin{array}{c} 2.16 \ m \\ 2.45 \ br \ dt \end{array} $	2.05 m }	2.04 m	2.02 m	2.05 m
H-14	6.48 br t	6.46 br t	5.10 br t	5.09 br t	5.09 br t	5.08 br t
H-16	9.36 s	9.39 s	1.68 br s	1.68 br s	1.69 br s	1.67 br s
H-17	1.73 br s	$1.75 \ br \ s$)		1.60 br s	1.59 br s
H-18	1.63 br s	$1.62 \ br \ s$	1.60 br s	1.60 br s	1.60 br s	1.60 br s
H-19	4.04 br s	10.00 s	{		1.65 br s	1.70 br s
H-20	4.05 br s	9.67 s	2.18 d	1.98 d	2.17 d	1.96 br s

Table 1. ¹H NMR spectral data of compounds 2–7 (400 MHz, CDCl₃, TMS as internal standard)

J(Hz): compounds: 2/3: 1,2 = 4,5 = 5,6 = 8,9 = 9,10 = 12,13 = 13,14 \sim 7; compounds 4/5: 1,2 = 8; 2,20 = 1; 4,5 = 5,6 = 9,10 = 13,14 = 7; compound 6: 1,2 = 8; 2,4 = 2,20 = 1; 4,5 = 8; 4,5' = 4.5; 5,5' = 14; 5,6 = 9,10 = 13,14 = 7; compound 7: 1,2 = 1.5; 5,6 = 9,10 = 13,14 = 7.

Table 2.	¹ H NM	IR spectral	data of	compounds	17b,	18b,	19	and	20	(CDCl ₃ ,
		400 M	Hz. TMS	S as internal s	standa	rd)				

	17 b	18Ь	19	20a	$20a + Eu(fod)_3$	20b
H-13	-				3.12 br s	
H-14 \	2.03 br d	2.01 br d	2.02 br d	2.05 dd	2.44 br d	2.05 dd
H-15					2.89 br d	
H-15'					2.62 br d	
H-17	2.89 d	2.85 d	2.89 d	2.87 d	5.02 br s	2.85 d
H-17'	2.81 d	2.75 d	2.81 d	2.80 d	4.96 br s	2.76 d
H-18	1.19 s	$1.17 \ s$	1.00 s	1.02 s	1.21 s	$0.99 \ s$
H-19	_	_	9.75 d	0.80 s	$0.87 \ s$	$0.79 \ s$
H-20	$0.85 \ s$	$0.85 \ s$	$0.89 \ s$	$0.84 \ s$	0.90 s	$0.85 \ s$
OMe	3.65 s	3.65 s		_		

J(Hz): compound 17b: 13,14 = 1.5; 14,14' = 12; 15,15' = 14.5; 17,17' = 4.5 (18b and 20b: 5.5); compound 19: 5,19 = 1.5.

large genera, however, different degrees of morphological development may be the reason for this diversity in chemistry.

EXPERIMENTAL

The air-dried plant material was extracted with Et₂O-petrol (1:2) and the resulting extracts were separated by column chromatography (Si gel) and further by TLC (Si gel). Known compounds were identified by comparing the IR and ¹H NMR spectra with those of authentic material.

Mikania luetzelburgii (voucher RMK 8119). The roots (70 g) (150 g) afforded 1 mg γ -humulene, 1 mg 22, 6 mg 23 and 4 mg 24, while the aerial parts (300 g) gave 5 mg germacrene D, 60 mg γ -humulene, 2 mg squalene, 8 mg 1, 6 mg phytol and 38 mg 2 (Et₂O).

Mikania sessilifolia (voucher RMK 8104). The roots (20 g) afforded 200 mg 8, 6 mg 9, 6 mg 10, 2 mg 11, 60 mg 12, 1 mg 14 (Et₂O-petrol, 1:1), 3 mg 17a (Et₂O-petrol, 1:1), 0.2 mg 18a

(Et₂O-petrol, 1:1) (both isolated as their methyl esters) and 1.2 mg 19 (Et₂O-petrol, 1:3), while the aerial parts (500 g) gave 25 mg germacrene D, 5 mg α -humulene, 50 mg squalene, 6 mg 4 (Et₂O-petrol, 1:10), 2 mg 5 (Et₂O-petrol, 1:10), 8 mg 6 (Et₂O-petrol, 1:1), 12 mg 7 (Et₂O-petrol, 1:50), 20 mg 8, 6 mg 12 and 4 mg 13.

Mikania luetzelburgii (voucher RMK 8119). The roots (70 g) afforded 50 mg germacrene D, 80 mg 8, 25 mg 10, 18 mg 12, 40 mg 13, 12 mg 15, 30 mg 16, 2 mg 17a and 8 mg 21, while the aerial parts (100 g) gave 10 mg germacrene D, 90 mg lupeyl acetate, 80 mg 8, 5 mg 10, 1 mg 12, 1 mg 13, 8 mg 15, 40 mg 16 and 5 mg 21.

Mikania belemii (voucher RMK 8007). The aerial parts (300 g) gave 50 mg α -humulene, 20 mg caryophyllene, 20 mg caryophyllene 5,6-epoxide, 5 mg gurjunene, 200 mg **8**, 100 mg **10**, 50 mg **12**, 5 mg **13** and 10 mg **16**.

19,20-Dihydroxy-16-oxo-geranyl nerol (2). Colourless gum, IR $\nu_{\rm max}^{\rm CHCl_3}$ cm $^{-1}$: 3600, 3400 (OH), 2730, 1680 (C=CCHO); MS m/z (rel. int.): 308 [M - CO] $^+$ (0.2), 235.170 [M - CH $_2$ C-

OHC OHC
$$\frac{16}{15}$$
 $\frac{14}{13}$ $\frac{12}{11}$ $\frac{10}{10}$ $\frac{8}{19}$ $\frac{4}{19}$ $\frac{3}{19}$ $\frac{1}{19}$ $\frac{1}{19}$

R = OH

4 R = H5 R = H(2.3Z)

10 11 12 13 16 R CO2H CO2H CO2H CH2OH CHO CO₂CH₂CH₂OH CO₂H CO₂H R' H OiVal OCinn OCinn H Н Н OH **OMeacr** R" H H OH H H H Н H

17a 17b 19 18a 18b 20a 20b CO₂H CO₂Me CO₂H CO₂Me CHO Me Me 16β-epoxide 16α-epoxide 16β-epoxide 16*B*

ÒR $22 \quad R = H$

23 R = iBu

OMe OAc ÒiBu 24

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 $(CH_2OH)=CHCH_2OH]^+$ (4) $(C_{15}H_{23}O_2)$, 124 (61), 55 (100); CI (isobutane): 337 $[M + 1]^+$ (6), 319 $[337 - H_2O]^+$ (28), 301 $[319 - H_2O]^+$ (100), 283 $[301 - H_2O]^+$ (58), 253 $[283 - CH_2O]^+$ (33). 10 mg 2 in 2 ml CHCl₃ were stirred 12 hr with 300 mg MnO₂. TLC (Et₂O-petrol, 1:1) afforded 6 mg 3, colourless gum; MS m/z (rel. int.): 247.133 [M - OCH(Me)= CHCH₂]⁺ (6) (C₁₅H₁₉O₃), 299 [247 – H₂O]⁺ (5), 55 (100). Geranylgeranal (4). Colourless oil, IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm⁻¹: 2740, 1685,

1618 (C=CCHO), 1640 (C=C); MS m/z (rel. int.): 288.245 [M]+ (6) $(C_{20}H_{32}O)$, 69 $[C_5H_9]^+$ (100).

Geranyl neral (5). Colourless oil, IR $v_{\text{max}}^{\text{CCl}_4}$ cm⁻¹: 2740, 1686, 1620 (C=CCHO); MS m/z (rel. int.): 288.245 [M]+ (6) $(C_{20}H_{32}O)$, 204 $[M - MeC(=CH_2)CH_2CHO]^+$ (4), 136 $[C_{10}H_{16}]^+$ (13), 84 $[C_5H_8O]^+$ (32), 69 $[C_5H_9]^+$ (100). 4-Hydroxygeranylgeranial (6). Colourless gum, IR $v_{max}^{CCl_4}$ cm⁻¹:

3600 (OH), 2740, 1740, 1680, 1615 (C=CCHO); MS m/z (rel. int.):

 $304.240 \, [M]^+ (1.5) (C_{20} H_{32} O_2), 286 \, [M - H_2 O]^+ (2.6), 100.052$ $[C_5H_8O_2]^+$ (30), 69 $[C_5H_9]^+$ (100).

Mikanifuran (7). Colourless oil, IR $v_{max}^{CCl_4}$ cm⁻¹: 1505, 895 (furan); MS m/z (rel. int.): 286.230 [M]⁺ (16) (C₂₀H₃₀O), 191.

$$[M - H_2\dot{C}]^+$$
 (9), 149 $[M - C_{10}H_{17}]^+$ (22), 95 $[C_5H_9]^+$ (100).

Methyl ent-kauranoate 16β,17-epoxide (17b). Colourless gum, IR $v_{\text{max}}^{\text{CCL}_4}$ cm⁻¹: 1730 (CO₂R); MS m/z (rel. int.): 332.235 [M]⁺ (18) (C₂₁H₃₂O₃), 300 [M - MeOH]⁺ (7), 273 [M - CO₂Me]⁺ (80), 55 $[C_4H_7]^+$ (100).

$$[\alpha]_{24}^{\lambda} = \frac{589}{-84} \frac{578}{-88} \frac{546}{-100} \frac{436 \text{ nm}}{-167} (c = 1.16, \text{ CHCl}_3).$$

Identical with the epoxide obtained by epoxidation of methyl ent-kauranoate (ca 90%).

Methyl ent-kauranoate-16α,17-epoxide (18b). Colourless gum, IR $v_{\text{max}}^{\text{CC14}}$ cm⁻¹: 1730 (CO₂R); MS m/z (rel. int.): 332.235 [M]⁺ (20) (C₂₁H₃₂O₃), 300 [M – MeOH]⁺ (4), 273 [M – CO₂Me]⁺ (100), identical with the epoxide from methyl ent-kaurenoate (ca 10 %).

Ent-kauran-19-al-16 β ,17-epoxide (19). Colourless gum, IR $V_{\rm max}^{\rm CCI_4}$ cm $^{-1}$: 2720, 1720, (CHO); MS m/z (rel. int.): 302.225 [M] $^+$ (41) (C₂₀H₃₀O₂), 273 [M - CHO] $^+$ (36), 255 [273 - H₂O] $^+$ (15), 55 [C₄H₇] $^+$ (100).

$$[\alpha]_{24}^{\lambda} = \frac{589}{-65} \quad \frac{578}{-71} \quad \frac{546}{-79} \quad \frac{436}{-123} \text{ ($c = 0.1$, CHCl}_3$).}$$

Preparation of 20a and 20b. 37 mg ent-kaurene in 2.5 ml $CHCl_3$ was stirred with 1.25 ml H_2O_2 and 1.25 ml HCO_2H for 5 hr at room temp. (see [20]). TLC afforded 5.2 mg 20a and 0.2 mg 20b.

20a:
$$[\alpha]_{24}^{\lambda} = \frac{589}{-69} \frac{578}{-72} \frac{546}{-82} \frac{436 \text{ nm}}{-140} (c = 0.89, \text{CHCl}_3).$$

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